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INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(51) International Patent Classification 6:
C09D 123/04

(11) International Publication Number: WO 97/03139
(43) International Publication Date: 30 January 1997 (30.01.97)

(21) International Application Number: PCT/FI96/00405

(22) International Filing Date: 10 July 1996 (10.07.96)

(30) Priority Data:

9502508-6 10 July 1995 (10.07.95) SE 962366 7 June 1996 (07.06.96) FI

(71) Applicant (for all designated States except US): BOREALIS POLYMERS OY [FI/FI]; P.O. Box 330, FIN-06101 Porvoo

(72) Inventors; and

(75) Inventors'Applicants (for US only): LEIDEN, Leif [FI/FI];
Norrveckkoski, FIN-07130 Anttila (FI). ASUMALAHTI,
Markku [FI/FI]; Louhenkuja 3, FIN-04230 Kerava (FI).
ÄÄRILÄ, Jari [FI/FI]; Partiomiehentie 1 B 31, FIN-06100
Porvoo (FI). ROGERSTEDT, Laila [SE/SE]; Näs 13025, S444 96 Ödsmål (SE). MARTINSSON, Hans-Bertil [SE/SE];
Ekbacken 9, S-472 95 Varekil (SE). HAGSTRÖM, Bengt
[SE/SE]; Tvetgatan 153, S-442 33 Kungalv (SE). SAHILA,
Aimo [FI/FI]; Tursontie 4 B 7, FIN-04200 Kerava (FI).

(74) Agent: BERGGREN OY AB; P.O. Box 16, FIN-00101 Helsinki (FI).

(81) Designated States: AL, AM, AT, AU, AZ, BB, BG, BR, BY, CA, CH, CN, CZ, DE, DK, EE, ES, FI, GB, GE, HU, IL, IS, JP, KE, KG, KP, KR, KZ, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, TJ, TM, TR, TT, UA, UG, US, UZ, VN, Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE).

Published

With international search report.

(54) Title: COATING COMPOSITION

(57) Abstract

Conventional coating compositions have the disadvantage of being insufficiently coatable and having an unsatisfactory environmental stress cracking resistance. Now a coating composition has been found which has both good coating processability and environmental stress cracking resistance. It comprises a multimodal ethylene polymer, which contains from 80 to 100 % by weight of ethylene repeating units and from 0 to 20 % by weight of C_3 - C_{10} α -olefin repeating units, has a density of between 0.915 g/cm³ and 0.955 g/cm³, and is a blend of at least a first ethylene polymer having a first average molecular weight and a first molecular weight distribution and a second ethylene polymer having a second average molecular weight, which is higher than said first average molecular weight, and a second molecular weight distribution, said blend having a third average molecular weight and a third molecular weight distribution.

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Coating composition

The invention relates to a coating composition and its use for coating a solid substrate. Coatings are applied to surfaces of all types to provide protection and decoration. Protection may be required against corrosion, oxidative ageing, weathering, or against mechanical damage.

When coating without solvent, the coating material should have good processing properties, i.e. the material should be easily melt coatable within a wide temperature interval, low shrinking, high mechanical strength, high surface finish and high environmental stress cracking resistance (ESCR). Since all these requirements have been difficult to fullfill, prior known coating materials have meant compromises whereby good properties in one sense have been achieved at the expense of good properties in another sense.

It would mean considerable advantages if the above mentioned compromise regarding the properties of a coating composition could be avoided. It would be particularly desirable to improve the coatability such as the melt flow properties when coating and the shrinkability of the coating material as well as the environmental stress cracking resistance of a product made from the coating material.

The purpose of the invention is to provide a coating material having good melt coating processability, low shrinking, high service temperature range and good environmental stress cracking resistance. The invention also strives for efficient coating speed expressed as high winding speed of the extruded material.

These purposes of the invention have now been achieved by means of a coating composition, which is substantially characterized in that it comprises a multimodal ethylene polymer, which contains from 80 to 100% by weight of ethylene repeating units and from 0 to 20% by weight of C₃-C₁₀ α-olefin repeating units, has a density of between 0.915 g/cm³ and 0.955 g/cm³, and is a blend of at least a first ethylene polymer having a first average molecular weight and a first molecular weight distribution and a second ethylene polymer having a second molecular weight, which is higher than said first molecular weight, and a second molecular weight distribution, said blend having a third molecular weight and a third molecular weight distribution.

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By a multimodal ethylene polymer is in connection with the present invention meant an ethylene polymer having broad molecular weight distribution produced by blending two or more ethylene polymer components with different molecular weights or by polymerizing ethylene to different molecular weights in a process with two or more reactors in series. By contrast, a unimodal ethylene polymer is obtained from only one ethylene polymer component produced in only one step.

The average molecular weights and the molecular weight distributions can be measured and expressed by any conventional method applied to ethylene polymer products. In this connection, it is convenient that the average molecular weights are measured and expressed as melt flow rates MFRⁱ_m, where i refers to said first, second and third average molecular weights and m refers to the load of the piston used for measuring the MFRs, which load in the following examples generally is 5.0 kg (m=5, see ISO 1133). The molecular weight distributions are conveniently expressed as flow rate ratios, FRRⁱ_m1/m², i.e. the ratios between high load MFRⁱs and low load MFRⁱs, where i refers to said first, second and third molecular weight distributions, and m¹ and m² refer to the high load, generally 21.6 kg (m=21), and low load, generally 5.0 kg (m=5) or 2.16 kg (m=2), respectively.

By Melt Flow Rate (MFR) is meant the weight of a polymer pressed through a standard cylindrical die at a standard temperature in a laboratory rheometer carrying a standard piston and load. Thus MFR is a measure of the melt viscosity of a polymer and hence also of its average molecular weight. The smaller the MFR, the larger is the average molecular weight. It is frequently used for characterizing a polyolefin, especially polyethylene, when the standard conditions MFR_m are: temperature 190°C; die dimensions 9.00 cm in length and 2.095 cm in diameter; load of the piston, 2.16 kg (m=2), 5.0 kg (m=5), 10.0 kg (m=10), 21.6 kg (m=21). See Alger, M.S.M., Polymer Science Dictionary, Elsevier 1990, p. 257.

By Flow Rate Ratio (FRR_m1/_m2) is meant the ratio between the melt flow rate (MFR_m1) measured at a standard temperature and with standard die dimensions using a heavy load (_m1) and the melt flow rate (MFR_m2) measured at the same temperature with the same die dimensions using a light load (_m2). Usually, for ethylene polymers, the heavy load m¹ is 21.6 kg (m¹=21) and the light load m² is 5.0 kg (m²=5) or 2.16 kg (m²=2) (ISO 1133). The larger the value of the FRR_m1/_m2, the broader is the molecular weight distribution.

The present invention is based on the finding, that multimodal ethylene polymer has excellent coating application properties such as good processability and low shrinkage as well as superior environmental stress cracking resistance.

5 The coating composition according to the present invention is a multimodal ethylene polymer. The multimodal ethylene polymer is by definition a blend of at least two ethylene polymers having different molecular weights. According to an important embodiment of the present invention, said blend is the product of a polymerization process comprising at least two steps. In the process, said first ethylene polymer is prepared by polymerizing ethylene in the presence of a catalyst system in a first step 10 and said second polymer is prepared by polymerizing ethylene in the presence of a catalyst system in a second step. Said steps can be performed in any order, whereby the ethylene polymer of each step is present in the following step or steps. However, it is preferential that said blend is the product of said polymerization process. 15 wherein said first step is performed before said second step. This means, that first, an ethylene polymer having a lower average molecular weight is produced and then. in the presence of the lower molecular weight ethylene polymer, an ethylene polymer having a higher average molecular weight is produced.

20 The idea of the present invention can be realized with any kind of ethylene polymerization catalyst, such as a chromium catalyst, a Ziegler-Natta catalyst or a group 4 transition metallocene catalyst. According to one embodiment of the present invention, said blend forming the multimodal ethylene polymer is the product of a polymerization process in which said first step and/or said second step is performed 25 in the presence of a catalyst system comprising a procatalyst based on a tetravalent titanium compound, such as a TiCl₄/MgCl₂/optional inert carrier/optional internal electron donor procatalyst, and a cocatalyst based on an organoaluminum compound, preferentially a R3Al/optional external electron donor procatalyst wherein R is a C₁-C₁₀ alkyl. Typical catalyst systems are e.g. prepared according to WO91/12182 and WO95/35323 which are herewith included by reference. A 30 preferential single site polymerization catalyst system is that based on a group 4 (IUPAC 1990) metal metallocene and alumoxane.

When performing said polymerization process comprising at least two steps, one or more catalyst systems, which may be the same or different, can be used. It is preferential, if said blend is the product of said polymerization process, in which said catalyst system is added to said first step and the same catalyst system is used at least in said second step.

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The most convenient way to regulate the molecular weight during the multistep polymerization of the present invention is to use hydrogen, which acts as a chain-transfer agent by intervening in the insertion step of the polymerization mechanism. Hydrogen may be added in suitable amounts to any step of the multistep polymerization. However, it is preferential that in said first step a hydrogen amount is used, leading to a melt flow rate MFR ¹₂ of said first ethylene polymer of from 50 g/10 min. to 2000 g/min., most preferentially from 100 g/10 min. to 1000 g/10 min., provided that said first step is performed before said second step.

It is prior known to prepare multimodal and especially bimodal olefin polymers in two or more polymerization reactors in series. Such processes are exemplified by EP 040992, EP 041796, EP 022376 and WO92/12182 which are hereby included as reference concerning the preparation of the multimodal ethylene polymers for the claimed coating material. According to these references each of said polymerization steps can be performed in liquid phase, slurry or gas phase.

According to the present invention, it is preferential to perform said polymerization steps as a combination of slurry polymerization and gas phase polymerization. Preferentially said first step is a slurry polymerization and said second step a gas phase polymerization.

The slurry polymerization is preferentially performed in a so called loop reactor. The gas phase polymerization is performed in a gas phase reactor. The polymerization steps can optionally be preceded by a prepolymerization, whereby up to 20% by weight and preferentially 1-10% by weight of the total ethylene polymer amount is formed.

Above, the use of hydrogen to regulate the molecular weight of the ethylene polymers is mentioned. The properties of the ethylene polymers can also be modified by adding a small amount of α -olefin to any of said polymerization steps. According to one embodiment of the present invention, said first ethylene polymer has a C₃-C₁₀ α -olefin repeating unit content of from 0 to 10% by weight of said first ethylene polymer. Preferentially, in said second ethylene polymer, the C₃-C₁₀ α -olefin, such as 1-butene or 1-hexene, repeating unit content is from 1 to 25% by weight, most preferentially 2 to 15% by weight, of said second ethylene polymer.

When using only two steps in said polymerization process, the ratio between the first produced ethylene polymer having the MFR¹₂ defined above, and the second

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produced ethylene polymer, having a lower MFR, is between 20:80 and 80:20, preferentially between 20:80 and 60:40.

The above mentioned polymerization conditions of the different steps can be coordinated so that the blend produced is most suitable for the coating of a solid substrate. Thus, the following preferential properties of said multimodal ethylene polymer blend can be achieved.

According to a preferred embodiment of the present invention, the melt flow rate MFR³₂ of said blend is between 0.1 and 50 g/10 min., more preferably between 0.1 and 20 g/10 min. According to a preferred embodiment, the C₃-C₁₀ α-olefin repeating unit content of said blend is from 0.2 to 20% by weight, more preferably from 0.5 to 15% by weight, of said blend. According to a preferred embodiment, the flow rate ratio FRR³_{21/5} of said blend is between 10 and 50, more preferentially between 15 and 40. Thereby, it is advantageous if the molecular weight distribution corresponding to said flow rate ratio shows several peaks or a broad peak lacking small fractions of extremely low and extremely high molecular weight material.

As was described above, the multimodal ethylene polymer according to the invention can be produced in a polymerizing process having at least two steps leading to different average molecular weights. According to another important embodiment of the invention, the multimodal ethylene polymer can be produced by mixing at least two ethylene polymers having different average molecular weights. In the latter case, said blend is a mechanical mixture of at least said first ethylene polymer and said second ethylene polymer, preferentially a mechanical mixture of said first ethylene polymer and said second ethylene polymer.

When mixing two ethylene polymers of different average molecular weight, the mixing is generally a melt mixing in a melt processing apparatus like a compounder and an extruder. The product is then a mechanical melt mixture of at least said first ethylene polymer and said second ethylene polymer. Preferentially there are only two ethylene polymers so that the mechanical melt mixture is a mixture of said first ethylene polymer and said second ethylene polymer. The preferable ratio between said first ethylene polymer and said second ethylene polymer is between 20:80 and 80:20, most preferably between 20:80 and 60:40.

When feeding the first ethylene polymer to the mixing step, the melt flow ratio MFR¹₂ of the first ethylene polymer is preferentially from 50 to 2000 g/10 min.,

most preferentially from 100 to 1000 g/10 min. When feeding at least said second ethylene polymer to the mixing step, its or their melt flow ratio MFR², etc.₂₁ is preferentially from 0.05 to 50 g/10 min., most preferentially from 0.10 to 20 g/10 min.

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In its most wide scope, the present invention relates to a coating composition comprising any multimodal ethylene polymer. This means, that the different ethylene polymers which can be used can have the monomer composition of a homopolymer or a copolymer. Preferentially said first ethylene polymer has a C_3 - C_{10} α -olefin repeating unit content of 0.0 to 10% by weight, calculated from the weight of said ethylene polymer.

Usually at least one ethylene polymer component of said blend is an ethylene copolymer containing a small amount of another α -olefin. Preferentially, said second ethylene polymer has a C_3 - C_{10} α -olefin, preferentially 1-butene or 1-hexene, repeating unit content of from 1.0 to 25% by weight and most preferentially from 2.0 to 15% by weight. Typical other comonomers are 4-methyl-1-pentene and 1-octene. When a blend of more than two ethylene polymer components is used, the further ethylene polymer components can be either homopolymers or copolymers.

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Conclusively, in the embodiment where the coating composition multimodal ethylene polymer is prepared by mixing at least a first ethylene polymer and a second ethylene polymer, the proportion of the first ethylene polymer and the second, etc., ethylene polymer, the MFR¹, MFR², etc. of said ethylene polymers and C_3 - C_{10} α -olefin repeating unit content of said ethylene polymers are preferentially such that the MFR³₂ of the blend obtained is between 0.1 and 50 g/10 min., preferentially between 0.1 and 20 g/10 min. Correspondingly, but independently, the C_3 - C_{10} α -olefin repeating unit content of the said blend is from 0.2 to 20% by weight, preferentially from 0.5 to 15% by weight. The flow rate ratio FRR³_{21/5} of said blend is between 10 and 50, preferentially between 15 and 40.

lack mate 35 unit

The molecular weight distribution curve shows either several peaks or a broad peak lacking small fractions of extremely low and extremely high molecular weight material. It was found, that, although the melt flow ratios, the comonomer repeating unit content and the flow rate ratio were essentially the same as in known monomodal products, the mere fact that the ethylene polymer was multimodal, i.e. was the blend of different molecular weight fractions, rendered it totally superior e.g.

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with respect to the processability measured as winding speed and the environmental stress cracking resistance.

Above, a multimodal ethylene polymer suitable for a coating composition has been described, which is merely the product of multistep polymerization or mixing. The invention also relates to a coating composition, prepared by a combination of multistep polymerization and mixing, e.g. by polymerizing ethylene in two or more steps and mixing the product with one or more ethylene polymers. Also, the finished polymerization or mixing product can further be treated to modify its average molecular weight and molecular weight distribution.

According to one embodiment of the present invention said blend is a blend which has been treated in a tailoring step comprising heating, melt processing and subjecting of a multimodal ethylene polymer to controlled radical reactions to give a molecular weight, which is at least as high as the molecular weight of the untreated blend, and a molecular weight distribution, which is broader than the molecular weight distribution of the untreated blend.

Preferentially a said tailoring step leads to a relative decrease in the melt flow ratio MFR₅, -(MFR³'₅ - MFR³₅): MFR³₅, is from 5 to 100%, preferentially from 10 to 80%, wherein MFR³'₅ is the melt index of said blend after said tailoring step. The upper limits are not to be interpreted as limitations, but it has only a descriptive function, which is based on the experimental results obtained in connection with the present invention. Anyhow, it appears that the melt viscosity decreases by several tens of percent, which meant that the controlled free radical reactions essentially lead to the combination of radical fragments into larger ethylene polymer molecules than before the controlled free radical reactions.

Perhaps even more important is the effect of the tailoring step on the molecular weight distribution expressed as the flow rate ratio of the blend. According to one embodiment of the invention, the relative broadening of the molecular weight distribution expressed as +(FRR³'21/5-FRR³21/5):FRR³21/5 is from 5 to 100%, preferentially from 10 to 80%, wherein FRR³'21/5 is the flow rate ratio of said blend after said tailoring step.

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The free radical reactions of said tailoring step can be effected in many ways. Firstly, free radicals may be generated from initiators in diverse ways, among which thermal or photochemical intermolecular bond cleavage, redox reactions, and

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photochemical hydrogen abstraction are the most common, but other processes such as the use of γ -radiation or electron beams find application. Free radicals can also be generated by means of reaction of the ethylene polymer blend by means of thermal decomposition with or without the presence of oxygen. Thermal treatment is a suitable method, especially if unstabilized or partly stabilized polyethylene is used or if the used ethylene polymer is destabilized during the treatment.

One of the main obstacles for using ethylene polymers as coating materials has up to now been an unsatisfying environmental stress cracking resistance. Another obstacle has been the poor melt coating processability of ethylene polymers. In the present invention, the finding that multimodal ethylene polymer has a totally superior environmental stress cracking resistance and melt coating processability, has lead to a new coating material which is the technical and commercial consequence of said finding. Thus the coating composition of the present invention preferentially has an environmental stress cracking resistance (ESCR, F20) (ASTM 1693/A, 10% Igepal) of at least 100 h, more preferentially at least 500 h, yet more preferentially at least 1000 h and most preferentially 2000 h.

In principle the claimed coating composition is suitable for any solid substrate such as a particle, powder, a corn, a grain, a granula, granulate, an aggregate, a fiber, a film, a bladder, a blister, a coat, a coating, a cover, a diaphragm, a membrane, a skin, a septum, a serving, a foil, a web, a cloth, a fabric, a canvas, a textile, a tissue, a sheet, a board, a cardboard, a carton, a fiber board, a paperboard, a pasteboard, a disc, a laminate, a layer, a planchette, a plate, a slab, a slice, a spacer, a wafer, a tape, a belt, a lace, a ribbon, a slip, a string, a strip, a band, a rope, a filament, a file, a fillet, a thread, a wire, a cable wire, a wire cable, a wire rope, a yarn, a cable, a conduit, a cord, a duct, a line, a rope, a hawser, a body, a block, a part, a moulding, a workpiece, a fabricated shape, a form part, a formed piece, a mould shape, a shaped piece, a special casting, a special piece, a bar, a lever, a boom, a pole, a rod, a shaft, a shank, a spoke, a staff, an arm, a gripe, a helve, a pin, a rein, a spindle, a staff, a stalk, a stem, a tube, a hose, a hose pipe, a sleeve, a barrel, a canal, a pipe, a spout, a valve, a profile.

Preferentially the claimed coating composition is a coating material for a solid substrate made of metal such as iron, steel, noble metals, metal alloys, composition metals, hard metals, cintered metals, cermets, or from non-metals such as concrete, cement, mortar, plaster, stone, glass, porcelain, ceramics, refractory materials.

enamel, wood, bark, cork, paper and paperboard, textile, leather, rubber and caoutchouc, plastics, and bituminous materials.

Most preferentially the claimed coating composition is a coating material for a rigid solid substrate, preferentially a rigid pipe, a rigid pipe fitting or a rigid profile, most preferentially an iron or steel pipe, pipe fitting or profile. In more detail, such a pipe is an iron or steel pipe, having a primer like an epoxy lacquer covering the steel surface and a coupling agent like carboxy modified polyethylene covering said primer. The claimed coating composition is then attached to the layer of carboxy modified polyethylene.

In addition to the above described coating composition, the present invention also relates to the process for the preparation of said coating composition, thereby the claimed process is as described above.

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The invention also relates to the use of a coating composition according to the above description or prepared by the claimed process for the coating of a solid substrate such as a particle, powder, a corn, a grain, a granula, granulate, an aggregate, a fiber, a film, a bladder, a blister, a coat, a coating, a cover, a diaphragm, a membrane, a skin, a septum, a serving, a foil, a web, a cloth, a fabric, a canvas, a textile, a tissue, a sheet, a board, a cardboard, a carton, a fiber board, a paperboard, a pasteboard, a disc, a laminate, a layer, a planchette, a plate, a slab, a slice, a spacer, a wafer, a tape, a belt, a lace, a ribbon, a slip, a string, a strip, a band, a rope, a filament, a file, a fillet, a thread, a wire, a cable wire, a wire cable, a wire rope, a yarn, a cable, a conduit, a cord, a duct, a line, a rope, a hawser, a body, a block, a part, a moulding, a workpiece, a fabricated shape, a form part, a formed piece, a mould shape, a shaped piece, a special casting, a special piece, a bar, a lever, a boom, a pole, a rod, a shaft, a shank, a spoke, a staff, an arm, a gripe, a sleeve, a barrel, a canal, a pipe, a spout, a valve, a profile.

Preferentially the use is directed to the coating of a solid substrate made of metal such as iron, steel, noble metals, metal alloys, composition metals, hard metals, cintered metals, cermets, or of non-metals such as concrete, cement, mortar, plaster, stone, glass, porcelain, ceramics, refractory materials, enamel, wood, bark, cork, paper and paperboard, textile, leather, rubber and caoutchouc, plastics, and bituminous materials.

The use of the invention is most preferably directed to the coating a rigid solid substrate, preferentially a rigid pipe, a rigid pipe fitting or a rigid profile, most preferentially an iron or steel pipe, pipe fitting or profile. In the case of a coating a metal pipe like an iron or steel pipe with the coating composition according to the invention, the pipe is preferentially covered with a primer like an epoxy lacquer and the primer layer is then covered with a coupling agent layer like carboxy modified polyethylene, whereby the coating composition is deposited on said coupling agent layer.

In the following, a few examples are presented to illuminate the present invention. In the examples, bimodal polyethylene was prepared and tested. The preparations were as follows.

Example 1

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Bimodal polyethene #1 was polymerized with a Ziegler-Natta -type catalyst prepared according to FI 942945 in one loop and one gas phase reactor which were operated in series. Ethene was polymerized in the presence of hydrogen in the loop reactor resulting in MFR₂=468. Ethene was polymerized with 1-butene and hydrogen in the gas phase. The production rate split of reactors was 45%/55%. The final product MFR₂=1.3, FRR_{21/5}=18 and density 941 kg/m³.

Bimodal polyethene #2 was polymerized with a Ziegler-Natta -type catalyst in one loop and one gas phase reactor which are operated in series. Ethene was polymerized in presence of hydrogen in the loop reactor resulting in MFR₂=444. Ethene was polymerized with 1-butene and hydrogen in the gas phase reactor. The production rate split of reactors was 40%/60%. The final product MFR₂=1.3, FRR_{21/5}=16 and density 940 kg/m³.

30 Reference material was commercial low shrinkage material HE6066 of Borealis.

Example 2

Bimodal polyethene #3 was polymerized with a Ziegler-Natta -type catalyst prepared according to FI 942949 in one loop and one gas phase reactor which are operated in series. Ethene was polymerized in presence of hydrogen in the loop reactor resulting in MFR₂=492. Ethene was polymerized with 1-butene and

hydrogen in the gas phase reactor. The production rate split of reactors was 45%/55%. The final product MFR₂=0.4, FRR_{21/5}=21 and density 941 kg/m³.

Bimodal polyethene #4 was polymerized with a Ziegler-Natta -type catalyst in one loop and one gas phase reactor which are operated in series. Ethene was polymerized in presence of hydrogen in the loop reactor resulting in MFR₂=53. Ethene was polymerized with 1-butene and hydrogen in the gas phase reactor. The production rate split of reactors was 44%/56%. The final product MFR₂=0.3, FRR_{21/5}=17 and density 941 kg/m³.

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Reference material was commercial steel pipe coating material HE6060 of Borealis.

Example 3

Bimodal polyethene #5 was polymerized with a Ziegler-Natta -type catalyst prepared according to FI 942949 in one loop and one gas phase reactor which are operated in series. Ethene was polymerized in the presence of hydrogen in the loop reactor resulting in MFR₂=384. Ethene was polymerized with 1-butene and hydrogen in the gas phase reactor. The production rate split of reactors was 45%/55%. The final product MFR₂=0.5, FRR_{21/5}=19 and base resin density 944 kg/m³.

Bimodal polyethene #6 was polymerized with a Ziegler-Natta -type catalyst prepared according to FI 942949 in one loop and one gas phase reactor which are operated in series. The catalyst was prepolymerized before feeding into the loop reactor. The prepolymerization degree was 62 g/g. Ethene was polymerized in the presence of hydrogen in the loop reactor resulting in MFR₂=274. Ethene was polymerized with 1-butene and hydrogen in the gas phase reactor. The production rate split of loop and gas phase reactors was 48%/52%. The final product MFR₂=0.5, FRR_{21/5}=20 and base resin density 945 kg/m³.

Bimodal polyethene #7 was polymerized with a Ziegler-Natta -type catalyst prepared according to FI 942949 in one loop and one gas phase reactor which are operated in series. Ethene was polymerized in the presence of hydrogen and 1-butene in the loop reactor resulting in MFR₂=230 and density 943 kg/m³. Ethene was polymerized with 1-butene and hydrogen in the gas phase reactor. The production rate split of reactors was 43%/57%. The final product MFR₂=0.5, FRR_{21/5}=19 and base resin density 927 kg/m³.

The main end product properties of examples 1-2 are shown in tables 1 and 2. Table 1 shows a comparison between the coating materials #1 and #2 according to the invention and a commercial material (HE6066). As can be seen from table 1, the environmental stress cracking resistance and winding speed of the coating material according to the invention are outstanding comparing to corresponding conventional coating material.

<u>Table 1</u> Properties of bimodal polyethene materials compared to existing commercial materials

MATERIAL	Polyethene #1	Polyethene #2	HE6066
MFR ₂ (g/10 min.)	1.3	1.3	2.0
FRR _{21/5}	18	16	11
Mw			
Mw/Mn			
DENSITY (kg/m ³)	941	940	941 _{base resin}
ESCR, F20 (h)	>2000	>2000	106
Shrink., 100°C/24 h (%)1	0.45	0.6	1
Tensile at yield (MPa)	21	21	20
Winding speed, max (m/min.) ²	42	42	6

¹ Cable line 210°C and 75 m/min.

Table 2 shows a comparison between coating materials #3 and #4 according to the present invention and a corresponding commercial coating material (HE6060). As can be seen from table 2, the materials according to the present invention are

superior over conventional coating materials also at lower MFR₂ level.

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Demag coating line: screw Ø 45 mm, L/D=24, die width = 450 mm, die gap = 0,5 mm.

Table 2 Properties of bimodal polyethene materials compared to existing commercial materials, MFR₂ ~ 0.3 g/10 min

MATERIAL	Polyethene #3	Polyethene #4	HE6060
MFR ₂ (g/10 min.)	0.4	0.3	0.29
MFR ₅ (g/10 min.)	1.7	1.2	1.3
MFR ₂₁ (g/10 min.)	34	20	22.0
FRR _{21/5}	20	17	17
Mw			217 000
Mw/Mn			13.1
DENSITY (kg/m ³)	941	941	937.3 _{base resin}
ESCR, F20 (h)	>2000	>2000	
Shrink., 100°C/24 h (%)1	0.9	1.1	-
Tensile at yield (MPa)	21	23	
		·	
Winding speed, max (m/min.) ²	37	40	4-10

⁵ l Cable line 210°C and 75 m/min.

Processability behaviour of reference material HE6060 and Polyethene
#5, #6 and #7 were tested on Barmag Ø 60 mm, L/D extruder having
240 mm wide flat die with die gap 1.5 mm. Extruder screw speed was
85 rpm.

MATERIAL	Melt temp., °C	Current, A	Pressure, bar
HE6060 (ref.)	270	42	379
Polyethene #5	266	38	341
Polyethene #6	269	37	342
Polyethene #7	256	42	368

Demag coating line: screw Ø 45 mm, L/D=24, die width = 450 mm, die gap = 0,5 mm.

One can see that in spite of higher melt temperature, the power consumption and die pressure of reference material was high indicating worse processability compared to materials related in this invention.

A very important charasteristics of coating material is maximum winding speed of material and thus maximum output rate of the extruder. Table 4 shows that at least 67% higher line speed can be ahieved with materials of this invention compared to reference material HE6060. Note that further increase of winding speed from 25 m/min is limited by equipment not by break off of Polyethene #5, #6 and #7 films.

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Table 4 Maximum winding speed of example 3 materials on Barmag Ø 60 mm, L/D=24 extruder. Die width was 240 mm and die gap 1.5 mm. Cast film was applied on the Ø 150 mm epoxy/ahdesive precoated steel pipe.

MATERIAL	Maximum winding speed, m/min	Output, kg/m ³
HE6060, reference	15	36
Polyethene #5	≥ 25	≥ 51.8
Polyethene #6	≥ 25	≥ 50.9
Polyethene #7	≥ 25	≥ 54

Claims

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- 1. Coating composition, characterized in that it comprises a multimodal ethylene polymer, which contains from 80 to 100% by weight of ethylene repeating units and from 0 to 20% by weight of C₃-C₁₀ α-olefin repeating units, has a density of between 0.915 g/cm³ and 0.955 g/cm³, and is a blend of at least a first ethylene polymer having a first average molecular weight and a first molecular weight distribution and a second ethylene polymer having a second average molecular weight, which is higher than said first average molecular weight, and a second molecular weight distribution, said blend having a third average molecular weight and a third molecular weight distribution.
- 2. Coating composition according to claim 1, characterized in that said blend is the product of a polymerization process comprising at least two steps in which said first ethylene polymer is prepared by polymerizing ethylene in the presence of a catalyst system in a first step and said second ethylene polymer is prepared by polymerizing ethylene in the presence of a catalyst system in a second step, said steps being performed in any order, the ethylene polymer of each step being present in the following step or steps.
- 20 3. Coating composition according to claim 2, characterized in that said blend is the product of said polymerization process, wherein said first step is performed before said second step.
- 4. Coating composition according to claim 2 or 3, characterized in that said blend is the product of said polymerization process in which said first step and/or said second step is performed in the presence of a catalyst system comprising a procatalyst based on a tetravalent titanium compound, such as a TiCl₄/MgCl₂/optional inert carrier/optional internal electron donor procatalyst, and a cocatalyst based on an organoaluminum compound, preferentially a R₃Al/optional external electron donor procatalyst wherein R is a C₁-C₁₀ alkyl.
 - 5. Coating composition according to claim 2, 3 or 4, characterized in that said blend is the product of said polymerization process, in which said catalyst system is added to said first step and the same catalyst system is used at least in said second step.
 - 6. Coating composition according to any of claims 2 to 5, characterized in that said blend is the product of said polymerization process, in the first step of which an

amount of chain transfer agent, preferentially hydrogen, is used, which gives a melt flow rate MFR 12 for the first ethylene polymer of from 50 g/10 min. to 2000 g/min., preferentially from 100 g/10 min. to 1000 g/10 min., provided that step 1 is performed first.

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- 7. Coating composition according to any of claims 2 to 6, characterized in that said blend is the product of said polymerization process, in which said first step and said second step are performed as a slurry polymerization, gas phase polymerization or a combination thereof, preferentially said first step as a slurry polymerization and said second step as a gas phase polymerization.
- 8. Coating composition according to any of claims 2 to 7, characterized in that in said first ethylene polymer, the C_3 - C_{10} α -olefin repeating unit content is from 0 to 10% by weight of said first ethylene polymer.

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9. Coating composition according to any of claims 2 to 8, characterized in that in said second ethylene polymer, the C_3 - C_{10} α -olefin, preferentially 1-butene or 1-hexene, repeating unit content is from 1 to 25% by weight, preferentially from 2 to 15% by weight, of said second ethylene polymer.

- 10. Coating composition according to any of claims 2 to 9, characterized in that in said blend, the ratio between said first ethylene polymer and said second ethylene polymer is between 20:80 and 80:20, preferentially between 20:80 and 60:40.
- 25 11. Coating composition according to any of claims 2 to 10, characterized in that the melt flow rate MFR³₂ of said blend is between 0.1 and 50 g/10 min., preferentially between 0.1 and 20 g/10 min.
- 12. Coating composition according to any of claims 2 to 11, characterized in that the C₃-C₁₀ α-olefin repeating unit content of said blend is from 0.2 to 20% by weight, preferentially from 0.5 to 15% by weight, of said blend.
- 13. Coating composition according to any of claims 2 to 12, characterized in that the flow rate ratio FRR³21/5 of said blend is between 10 and 50, preferentially between 15 and 40, the molecular weight distribution curve showing several peaks or a broad peak lacking small fractions of extremely low and extremely high molecular weight material.

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14. Coating composition according to claim 1, characterized in that said blend is a mechanical mixture of at least said first ethylene polymer and said second ethylene polymer, preferentially a mechanical mixture of said first ethylene polymer and said second ethylene polymer.

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15. Coating composition according to claim 14, characterized in that said blend is a mechanical melt mixture of at least said first ethylene polymer and said second ethylene polymer, preferentially a mechanical melt mixture of said first ethylene polymer and said second ethylene polymer.

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- 16. Coating composition according to claim 14 or 15, characterized in that the ratio between said first ethylene polymer and said second ethylene polymer is between 20:80 and 80:20, preferentially between 20:80 and 60:40.
- 15 17. Coating composition according to claim 14, 15 or 16, characterized in that the melt flow ratio MFR¹₂ of said first ethylene polymer is from 50 to 2000 g/10 min., preferentially from 100 to 1000 g/10 min.
- 18. Coating composition according to any of claims 14 to 17, characterized in that the melt flow ratio MFR²₂₁ of said second ethylene polymer is from 0.05 to 50 g/10 min., preferentially from 0.1 to 20 g/10 min.
 - 19. Coating composition according to any of claims 14 to 18, characterized in that said first ethylene polymer has a C_3 - C_{10} α -olefin repeating unit content of from 0 to 10% by weight of said first ethylene polymer.
 - 20. Coating composition according to any of claims 14 to 19, characterized in that said second ethylene polymer has a C_3 - C_{10} α -olefin, preferentially 1-butene or 1-hexene, repeating unit content of from 1 to 25% by weight, preferentially from 2 to 15% by weight, of said second ethylene polymer.
 - 21. Coating composition according to any of claims 14 to 20, characterized in that the melt flow ratio MFR³₂ of said blend is between 0.1 and 50 g/10 min., preferentially between 0.1 and 20 g/10 min.

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22. Coating composition according to any of claims 14 to 21, characterized in that the C_3 - C_{10} α -olefin repeating unit content of said blend is from 0.2 to 20% by weight, preferentially from 0.5 to 15% by weight.

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- 23. Coating composition according to any of claims 14 to 22, characterized in that the flow rate ratio FRR³21/5 of said blend is between 10 and 50, preferentially between 15 and 40, the molecular weight distribution curve showing several peaks or a broad peak lacking small fractions of extremely low and extremely high molecular weight material.
- 24. Coating composition according to any previous claim, characterized in that said blend is a blend which has been treated in a tailoring step comprising heating, melt processing and subjecting to controlled free radical reactions to give a molecular weight, which is at least as high as the molecular weight of the untreated blend, and a molecular weight distribution, which is broader than the molecular weight distribution of the untreated blend.
- 25. Coating composition according to claim 24, characterized in that said blend is a blend which has been treated in said tailoring step so that during said tailoring step, the relative decrease in MFR₅, -(MFR³'₅ MFR³₅):MFR³₅, is from 5 to 100%, preferentially from 10 to 80%, wherein MFR³'₅ is the melt index of said blend after said tailoring step.
- 26. Coating composition according to claim 24, characterized in that said blend is a blend which has been treated in said tailoring step so that during said tailoring step, the relative broadening of the molecular weight distribution expressed as +(FRR³'21/5 FRR³21/5):FRR³21/5 is from 5 to 100%, preferentially from 10 to 80%, wherein FRR³'21/5 is the flow rate ratio of said blend after said tailoring step.
 - 27. Coating composition according to any previous claim, characterized in that it has a environmental stress cracking resistance (ESCR, F20) of at least 100 h, preferentially at least 500 h, more preferentially at least 1000 h and most preferentially at least 2000 h.
 - 28. Coating composition according to any previous claim, characterized in that it is a coating material for a solid substrate such as a particle, powder, a corn, a grain, a granula, granulate, an aggregate, a fiber, a film, a bladder, a blister, a coat, a coating, a cover, a diaphragm, a membrane, a skin, a septum, a serving, a foil, a web, a cloth, a fabric, a canvas, a textile, a tissue, a sheet, a board, a cardboard, a carton, a fiber board, a paperboard, a pasteboard, a disc, a laminate, a layer, a planchette, a plate, a slab, a slice, a spacer, a wafer, a tape, a belt, a lace, a ribbon, a slip, a string, a strip, a band, a rope, a filament, a file, a fillet, a thread, a wire, a

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cable wire, a wire cable, a wire rope, a yarn, a cable, a conduit, a cord, a duct, a line, a rope, a hawser, a body, a block, a part, a moulding, a workpiece, a fabricated shape, a form part, a formed piece, a mould shape, a shaped piece, a special casting, a special piece, a bar, a lever, a boom, a pole, a rod, a shaft, a shank, a spoke, a staff, an arm, a gripe, a helve, a pin, a rein, a spindle, a staff, a stalk, a stem, a tube, a hose, a hose pipe, a sleeve, a barrel, a canal, a pipe, a spout, a valve, a profile.

- 29. Coating composition according to any previous claim, characterized in that it is a coating material for a solid substrate made of metal such as iron, steel, noble metals, metal alloys, composition metals, hard metals, cintered metals, cermets, or from non-metals such as concrete, cement, mortar, plaster, stone, glass, porcelain, ceramics, refractory materials, enamel, wood, bark, cork, paper and paperboard, textile, leather, rubber and caoutchouc, plastics, and bituminous materials.
- 15 30. Coating composition according to any previous claim, **characterized** in that it is a coating material for a rigid solid substrate, preferentially a rigid pipe, a rigid pipe fitting or a rigid profile, most preferentially an iron or steel pipe, pipe fitting or profile.
- 20 31. Coating composition according to claim 30, characterized in that it is a coating material for a rigid solid substrate comprising an iron or steel pipe, a primer, like an epoxy laquer, covering said steel pipe, and a coupling agent, like carboxy modified polyethylene, covering said primer.
- 25 32. Use of a coating composition according to any preceding claim for the coating of a solid substrate such as a particle, powder, a corn, a grain, a granula, granulate, an aggregate, a fiber, a film, a bladder, a blister, a coat, a coating, a cover, a diaphragm, a membrane, a skin, a septum, a serving, a foil, a web, a cloth, a fabric. a canvas, a textile, a tissue, a sheet, a board, a cardboard, a carton, a fiber board, a paperboard, a pasteboard, a disc, a laminate, a layer, a planchette, a plate, a slab, a 30 slice, a spacer, a wafer, a tape, a belt, a lace, a ribbon, a slip, a string, a strip, a band, a rope, a filament, a file, a fillet, a thread, a wire, a cable wire, a wire cable, a wire rope, a yarn, a cable, a conduit, a cord, a duct, a line, a rope, a hawser, a body, a block, a part, a moulding, a workpiece, a fabricated shape, a form part, a formed 35 piece, a mould shape, a shaped piece, a special casting, a special piece, a bar, a lever, a boom, a pole, a rod, a shaft, a shank, a spoke, a staff, an arm, a gripe, a helve, a pin, a rein, a spindle, a staff, a stalk, a stem, a tube, a hose, a hose pipe, a sleeve, a barrel, a canal, a pipe, a spout, a valve, a profile.

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- 33. Use according to claim 32 for the coating of a solid substrate made of metal such as iron, steel, noble metals, metal alloys, composition metals, hard metals, cintered metals, cermets, or of non-metals such as concrete, cement, mortar, plaster, stone, glass, porcelain, ceramics, refractory materials, enamel, wood, bark, cork, paper and paperboard, textile, leather, rubber and caoutchouc, plastics, and bituminous materials.
- 34. Use according to claim 32 or 33 for the coating of a rigid solid substrate, preferentially a rigid pipe, a rigid pipe fitting or a rigid profile, most preferentially a iron or steel pipe, pipe fitting or profile.
- 35. Use according to claim 32, 33 or 34 for the coating of a rigid solid substrate comprising an iron or steel pipe, a primer, like an epoxy laquer, covering said steel pipe, and a coupling agent, like carboxy modified polyethylene, covering said primer.

INTERNATIONAL SEARCH REPORT

International application No. PCT/FI 96/00405

A. CLASSIFICATION OF SUBJECT MATTER

IPC6: C09D 123/04
According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC6: CO8L, CO9D

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

SE,DK,FI,NO classes as above

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

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C. DOCU	MENTS CONSIDERED TO BE RELEVANT	
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
х	GB 942369 A (E.I. DU PONT DE MEMOURS AND COMPANY), 20 November 1963 (20.11.63), page 1, line 30 - line 43, claims 1-3	1,14-16,18, 21,28-29, 32-33
		
x	EP 0022376 A2 (MITSUI PETROCHEMICAL INDUSTRIES, LTD.), 14 January 1981 (14.01.81), page 14, line 19 - line 21, claims 1-6	1-5,7-10,12, 28-29,32-33
х	US 4547551 A (FAY W. BAILEY ET AL), 15 October 1985 (15.10.85), claims 1-4, abstract, column 2, line 37-40, column 3, tabel I	1,14-23, 28-29,32-33
		

X	Further documents are listed in the continuation of Box	C.	X See patent family annex.
* "A" "E" "L"	Special categories of cited documents: document defining the general state of the art which is not considered to be of particular relevance erlier document but published on or after the international filing date document which may throw doubts on priority claim(s) or which is	"T"	later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention document of particular relevance: the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
"O"	cited to establish the publication date of another citation or other special reason (as specified) document referring to an oral disclosure, use, exhibition or other means document published prior to the international filing date but later than the priority date claimed	"Y"	document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art document member of the same patent family
	e of the actual completion of the international search October 1996	Date o	of mailing of the international search report 0 3 -10- 1996
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INTERNATIONAL SEARCH REPORT

International application No.
PCT/FI 96/00405

		PC1/F1 96/0	
C (Continu	ation). DOCUMENTS CONSIDERED TO BE RELEVANT	· · · · · · · · · · · · · · · · · ·	7
Category*	Citation of document, with indication, where appropriate, of the relevant	ant passages	Relevant to claim No.
Ρ,Χ	EP 0679704 A1 (SUMITOMO CHEMICAL COMPANY, LIMIT 2 November 1995 (02.11.95), page 1, line 1 - line 7; page 4, line 4 - line 17, abstract, claims 1-3, 7, 9-10, 21	1,14-16, 18-20,28-34	
			
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INTERNATIONAL SEARCH REPORT

Information on patent family members

05/09/96

International application No. PCT/FI 96/00405

Patent document cited in search report		Publication date	Patent family member(s)		Publication date	
GB-A-	942369	20/11/63	DE-B- FR-A- US-A-	1180935 1322615 3125548	00/00/00 00/00/00 00/00/00	
EP-A2-	0022376	14/01/81	JP-A- US-A-	56010506 4307209	03/02/81 22/12/81	
US-A-	4547551	15/10/85	CA-A- EP-A,B- SE-T3- JP-B- JP-A- US-A-	1216392 0100843 0100843 4029695 59047242 4461873	06/01/87 22/02/84 19/05/92 16/03/84 24/07/84	
EP-A1-	0679704	02/11/95	CA-A- CN-A- JP-A-	2146881 1117508 7292310	28/10/95 28/02/96 07/11/95	

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